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Die Veränderung des Volumens des Sauerstoffions in Alkali-Boratgläsern

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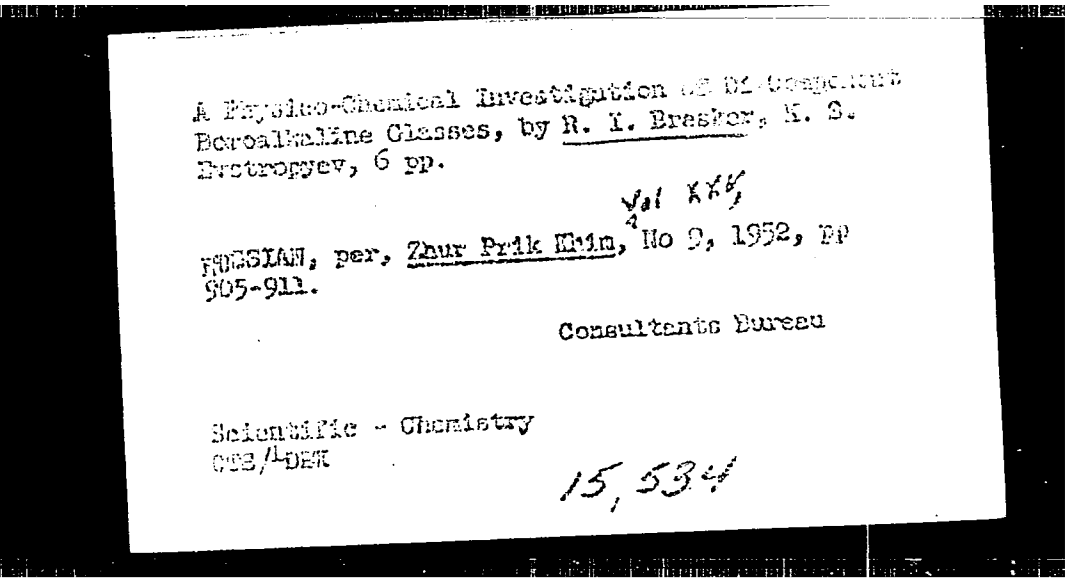

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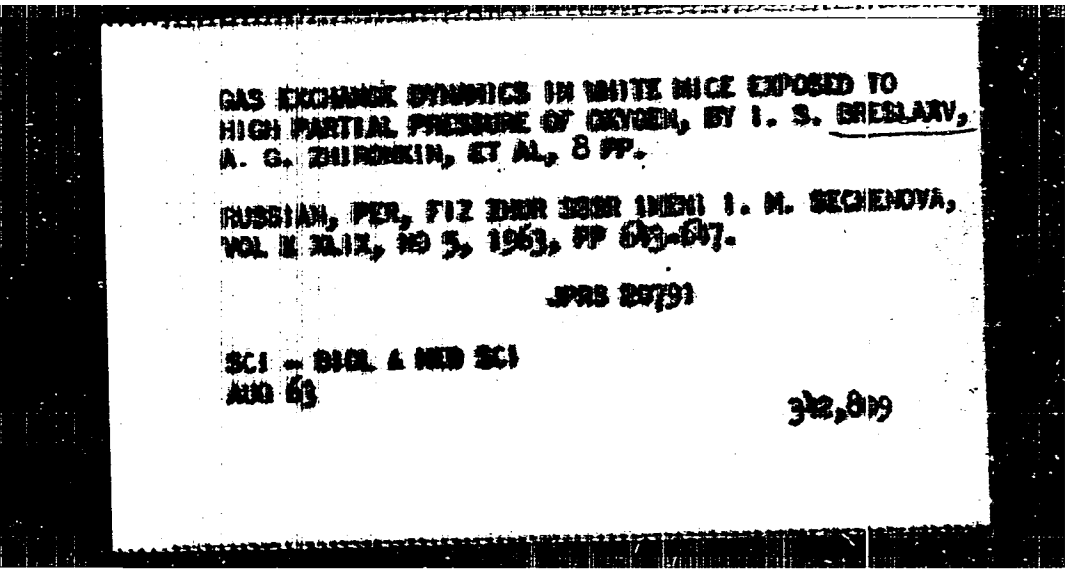

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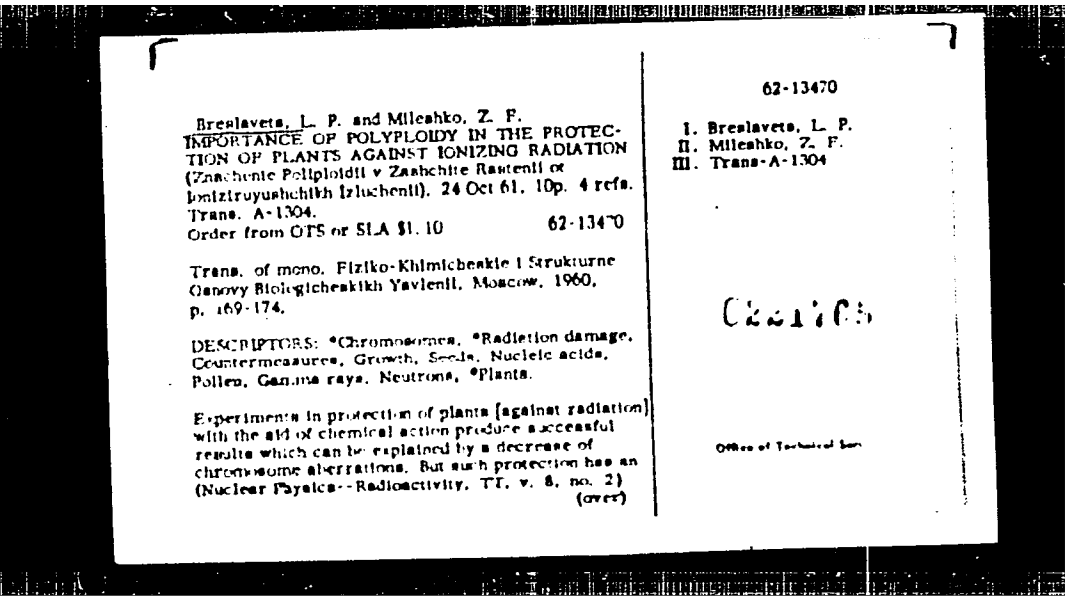

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Experiments in protection of plants [against radiation] with the aid of chemical action produce successful results which can be explained by a decrease of chromosome aberrations. But such protection has an
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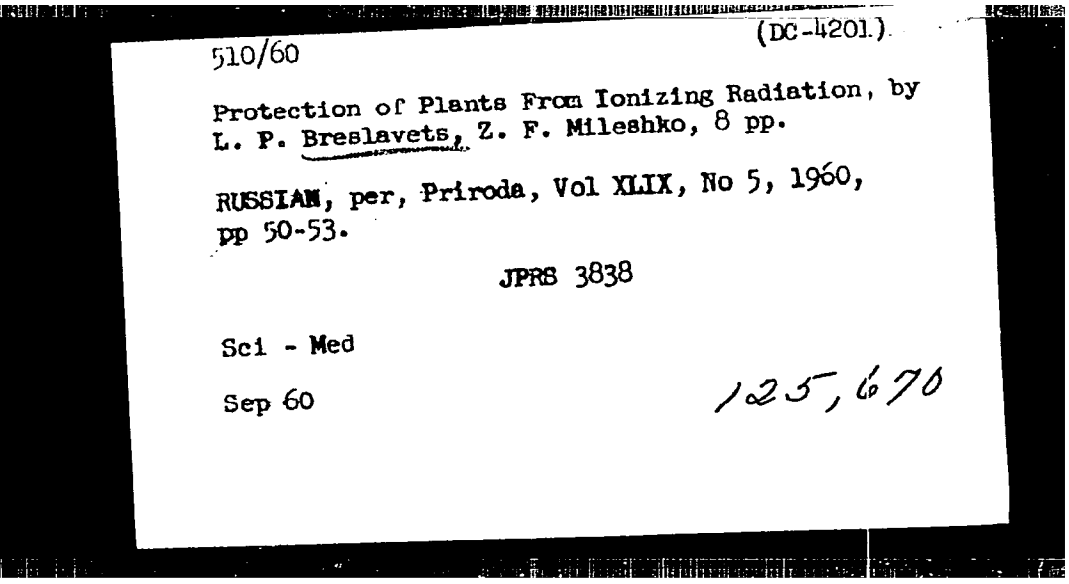

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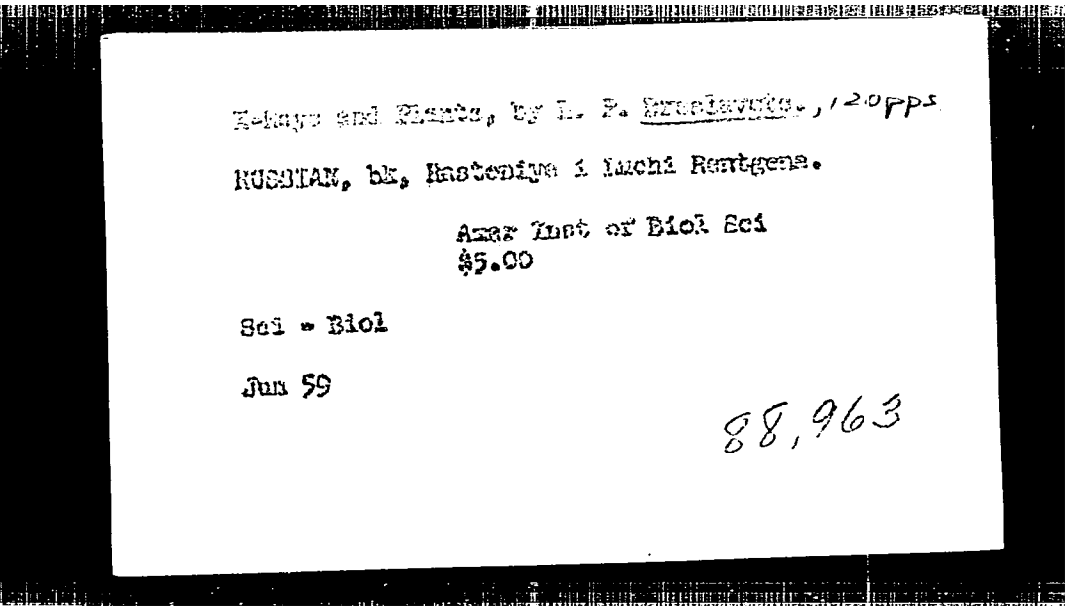

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Bresler, L. S., Dolgoplosk, B. A., Kolechkova, M. F., and Kropacheva, E. N.
COPOLYMERIZATION OF BUTADIENE WITH ISOPRENE UNDER THE INFLUENCE OF COMPLEXES OF BUTYLLITHIUM WITH TRIETHYLAMINE AND TETRAHYDROFURAN. [1963] 5p (table omitted) 11refs
Order from OTS or SLA $1.10 63-18852

Trans. of Akademiya Nauk SSSR. Doklady, 1962, v. 144, no. 2, p. 347-348.

DESCRIPTORS: *Butadienes, *Isoprene, *Copolymerization, Complex compounds, Amines, Methanes, Ethyl radicals, Butyl radicals, *Lithium compounds, Oxides, Synthetic rubber.

This paper contains the results of the study of the copolymerization of butadiene and isoprene with com- (Materials--Elastomers, TT, v. 10, no. 11) (over)

63-18852

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2. Title: Tetrahydrofuran
3. Title: Triethylamine
I. Bresler, L. S.
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The Mechanism of the Conversions of Tertiary Alcohols of the Cyclopropane Series under the Influence of Mineral and Organic Acids. VI. The Reaction of Methylethynylcyclopropylcarbinol with Sulfuric Acid and with Phosphorus Trichloride in the Presence of Pyridine. The Synthesis of Methyltrichloromethylcyclopropylcarbinol and the Investigation of its Reactions with Sulfuric Acid, by T. A. Favorskaya and L. S. Bresler, 6 pp.

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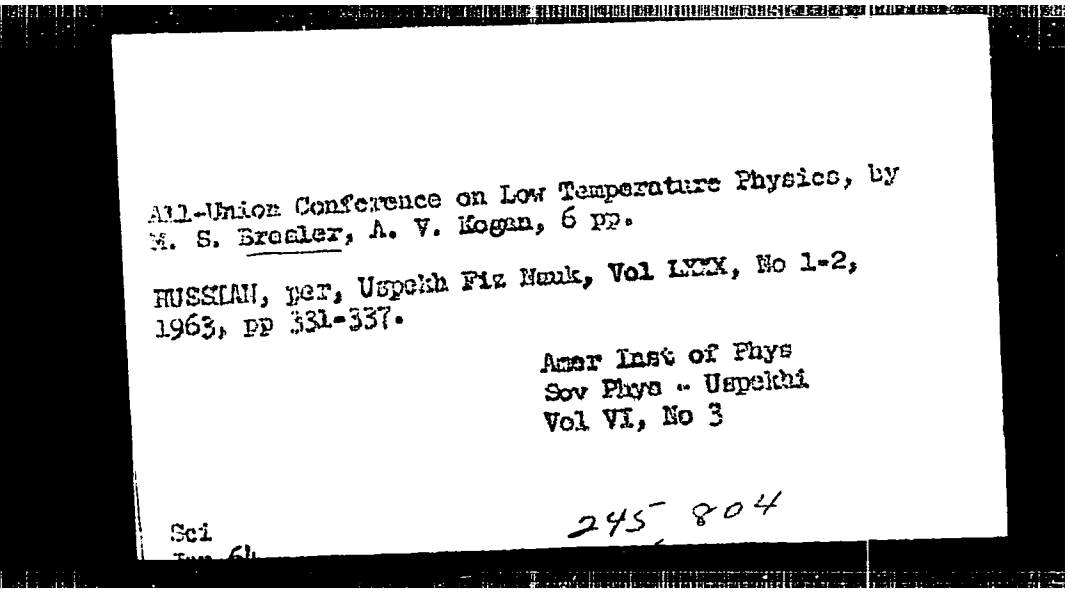

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DEVELOPMENTS IN THE SYNTHESIS AND UTILIZATION OF ION-EXCHANGE AND ELECTRON-EXCHANGE RESINS. [1961] 20p.
Order from ATS $26.95 ATS-52N49R

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(Chemistry--Organic, TT, v. 5, no. 11)

61-22130

1. Ion exchange resins--Synthesis
2. Ion exchange resins--Applications
3. Title: Electron exchange resins

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II. ATS-52N49R
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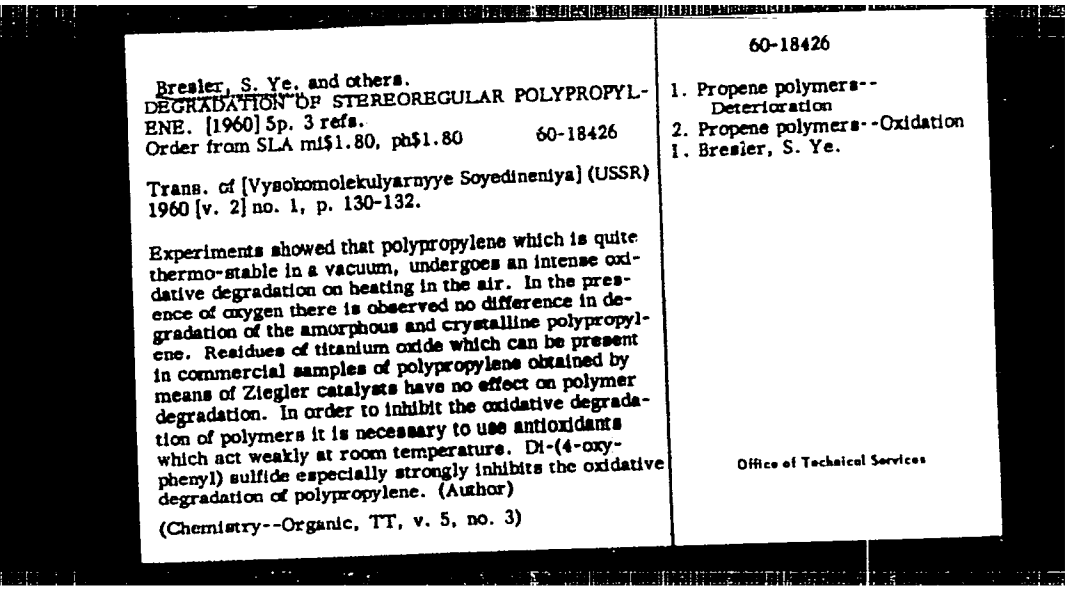

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Experiments showed that polypropylene which is quite thermo-stable in a vacuum, undergoes an intense oxidative degradation on heating in the air. In the presence of oxygen there is observed no difference in degradation of the amorphous and crystalline polypropylene. Residues of titanium oxide which can be present in commercial samples of polypropylene obtained by means of Ziegler catalysts have no effect on polymer degradation. In order to inhibit the oxidative degradation of polymers it is necessary to use antioxidants which act weakly at room temperature. Di-(4-oxyphenyl) sulfide especially strongly inhibits the oxidative degradation of polypropylene. (Author)

(Chemistry--Organic, TT, v. 5, no. 3)

60-18426

1. Propene polymers--Deterioration
2. Propene polymers--Oxidation
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Abstract of
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High Polymers Institute, Acad. Sci., USSR

CIA/FDD X-2731

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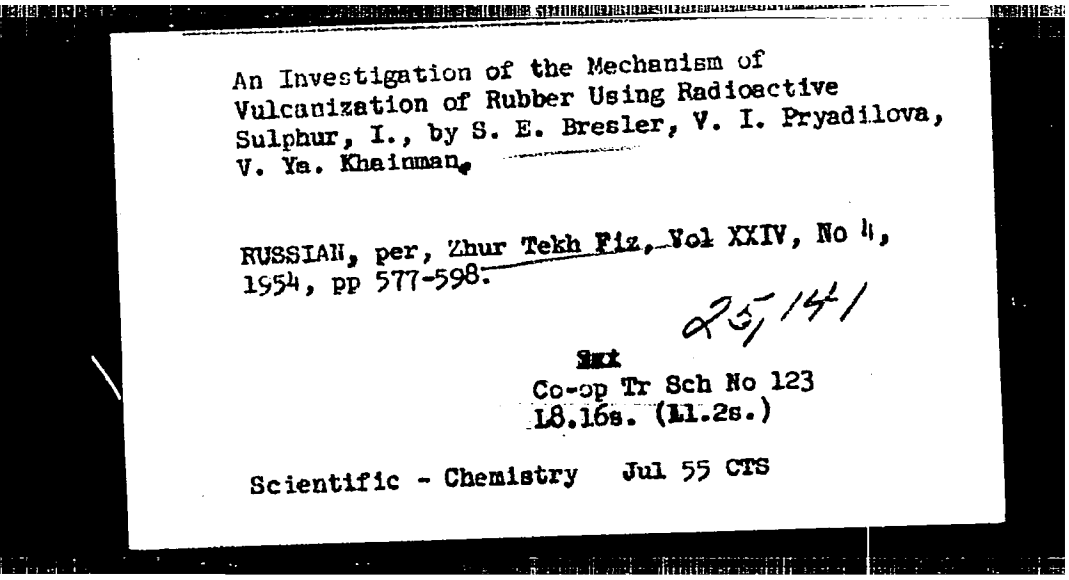

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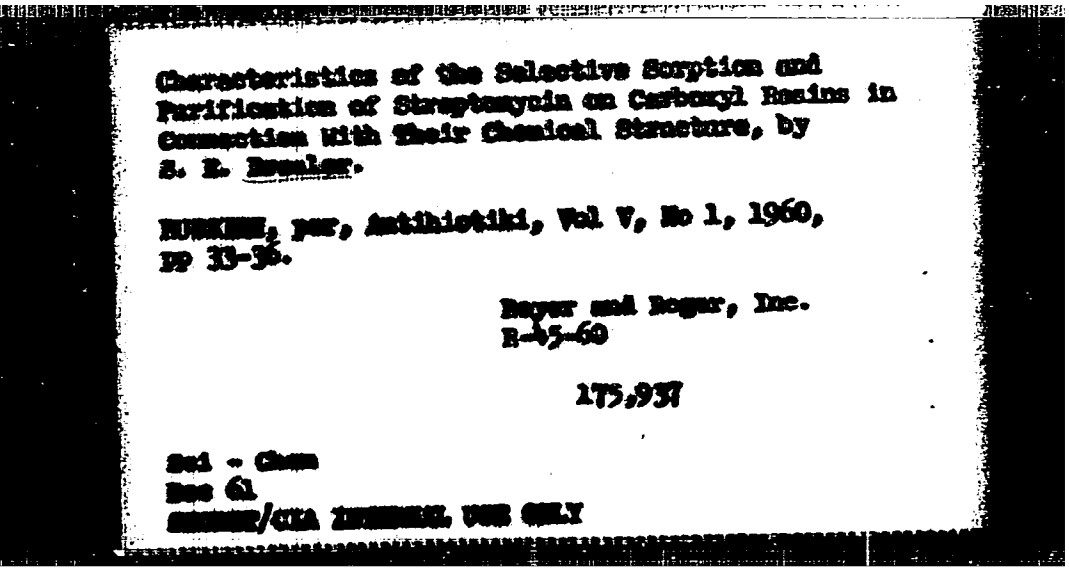

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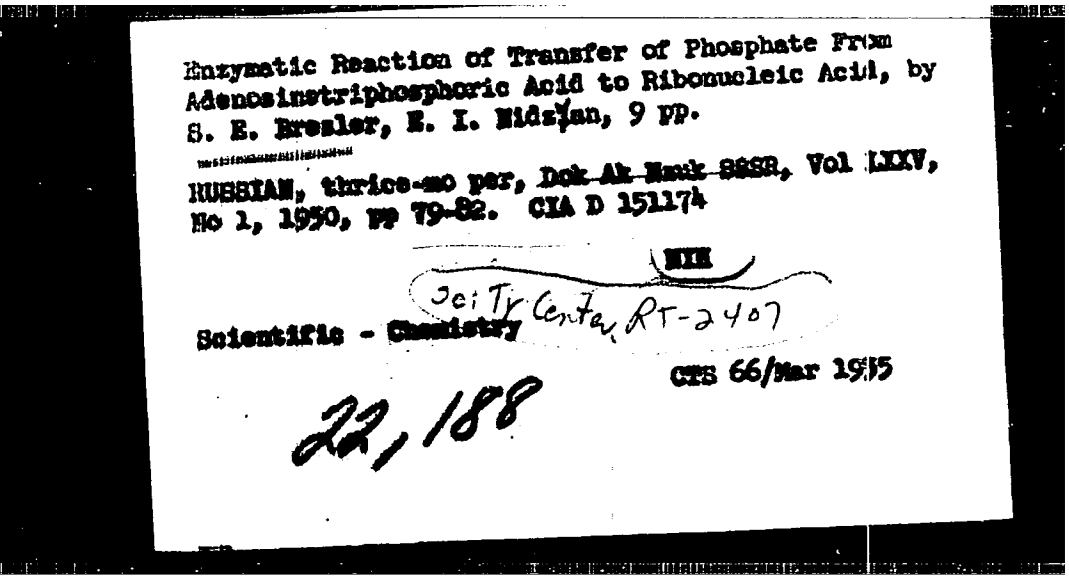

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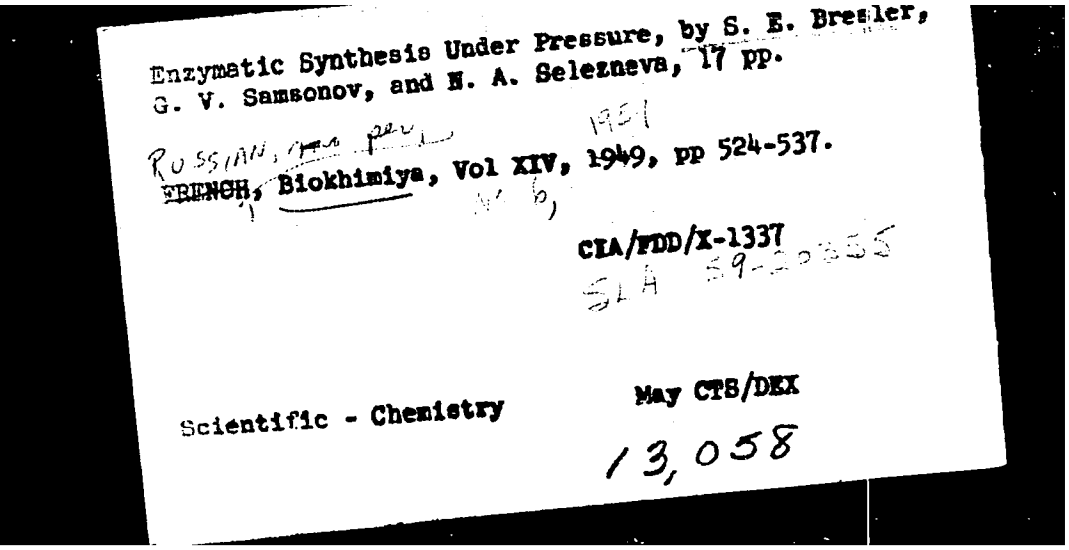

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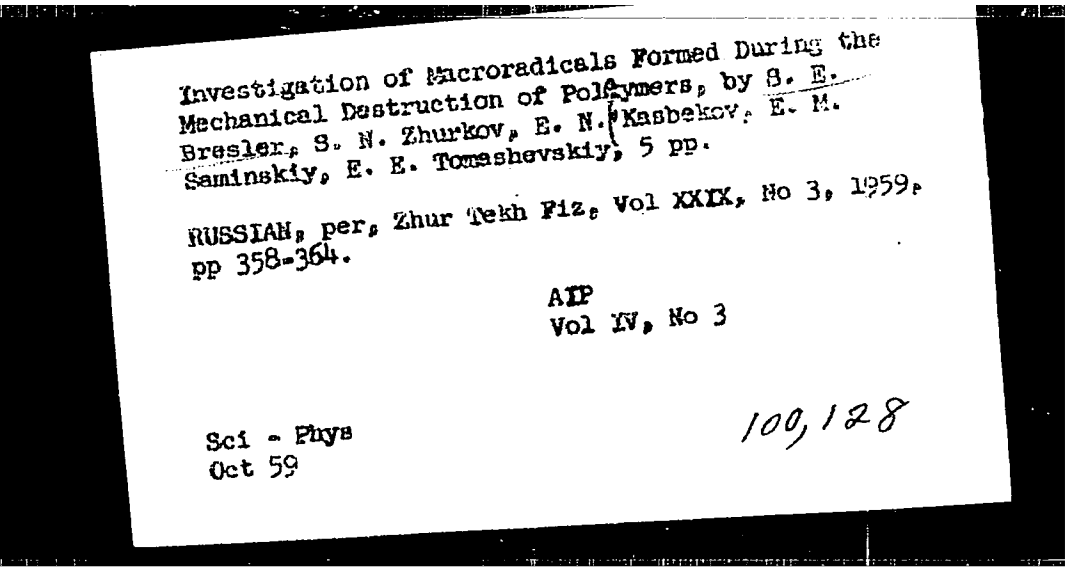

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Trans. of [Akademiya Nauk SSSR]. Comptes Rendus (Doklady) de l'Academie des Sciences de l'U. R. S. S., 1945, v. 47, p. 410-411.

DESCRIPTORS: *Hydrocarbons, *Polymerization, *Butenes, *Butadienes, Temperature, Polymers.

The autoacceleration of the polymerization of isobutene in the presence of boron fluoride is a chain process involving a phenomenon of explosion-like nature occurring at -100°C. Observations on the polymerization of 1, 3-butadiene and of isoprene with boron fluoride show that a critical initial temperature exists which is the exact equivalent of the lower limit of the chain explosions. For butadiene the critical temperature is -76°C
(Chemistry--Organic, TT, v. 6, no. 7) (over)

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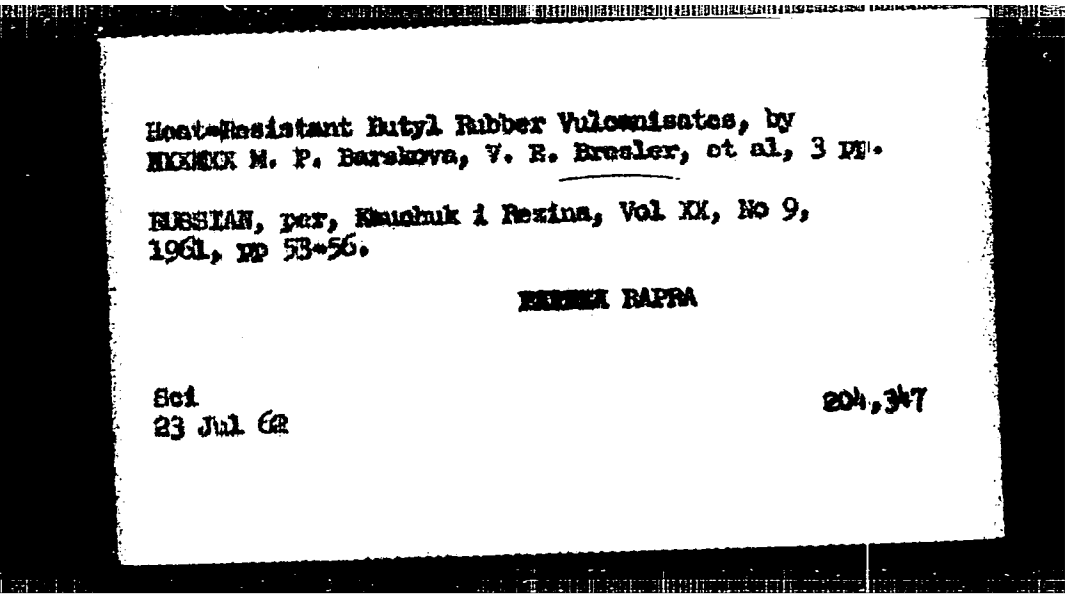

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